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INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

(Chapter II of the Patent Cooperation Treaty)

(PCT Article 36 and Rule 70)

Applicant's or agent's file reference P60188PCT		FOR FURTHER A	ACTION	See Form PCT/IPEA/416			
International application No. PCT/EP2004/012851		International filing date 09.11.2004	(day/month/year)	Priority date (day/month/year) 19.11.2003			
International Patent Classification (IPC) or national classification and IPC C07D241/46, C25D3/38							
Applicant ATOTECH DEUTS	Applicant ATOTECH DEUTSCHLAND GMBH et al						
This report is the Authority under	 This report is the international preliminary examination report, established by this International Preliminary Examining Authority under Article 35 and transmitted to the applicant according to Article 36. 						
2. This REPORT							
3. This report is al	This report is also accompanied by ANNEXES, comprising:						
a. 🛭 sent to t	he applicant and to	the International Bur	eau) a total of 12 sheet	s, as follows:			
⊠ shee and/ Adm	ets of the description for sheets containing inistrative Instructi	on, claims and/or drawing rectifications authorons).	ings which have been a ized by this Authority (so	mended and are the basis of this report ee Rule 70.16 and Section 607 of the			
Sup	plemental Box.	m the international ap	Jiication as filed, as indi	iders contain an amendment that goes cated in item 4 of Box No. I and the			
			ndicate type and numbe computer readable form 02 of the Administrative	er of electronic carrier(s)) , containing a only, as indicated in the Supplemental Instructions).			
4. This report cont	ains indications rel	ating to the following i	tems:				
☑ Box No. I	Basis of the opin	ion					
☐ Box No. II	Priority ·						
☐ Box No. III	Non-establishme	nt of opinion with rega	ard to novelty, inventive	step and industrial applicability			
☐ Box No. IV	Lack of unity of in	nvention					
⊠ Box No. V	applicability, cital	ions and explanations	 with regard to novelty, supporting such statem 	, inventive step or industrial nent			
☐ Box No. VI	Certain documen						
☐ Box No. VII		the international app					
☐ Box No. VIII	Certain observati	ons on the internation	al application				
Date of submission of the demand			Date of completion of this	s report			
28.04.2005			08.12.2005				
Name and mailing address of the international preliminary examining authority:			Authorized Officer	under Peterson			
European Patent Office - P.B. 5818 Patentlaan 2 NL-2280 HV Rijswijk - Pays Bas Tel. +31 70 340 - 2040 Tx: 31 651 epo nl Fax: +31 70 340 - 3016			Allard, M Telephone No. +31 70 34	10-2002			

INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

International application No. PCT/EP2004/012851

_		
_	Box No. I Bas	sis of the report
1.	With regard to t filed, unless oth	ne language , this report is based on the international application in the language in which it was
	which is the	is based on translations from the original language into the following language , language of a translation furnished for the purposes of:
	publicat	onal search (under Rules 12.3 and 23.1(b)) on of the international application (under Rule 12.4) onal preliminary examination (under Rules 55.2 and/or 55.3)
2.	nave been turni	ne elements* of the international application, this report is based on <i>(replacement sheets which</i> shed to the receiving Office in response to an invitation under Article 14 are referred to in this ally filed" and are not annexed to this report):
	Description, Pag	es
	1-4, 6-27	as originally filed
	5, 5a	received on 28.04.2005 with letter of 22.04.2005
	Claims, Numbers	•
	1-29	received on 28.04.2005 with letter of 22.04.2005
	☐ a sequence	listing and/or any related table(s) - see Supplemental Box Relating to Sequence Listing
3.	☐ The amend	ments have resulted in the cancellation of:
		ription, pages
	☐ the clain☐ the draw	is, Nos. ings, sheets/figs
	🗆 the sequ	ence listing (specify):
	☐ any table	e(s) related to sequence listing (specify):
4.	nad not been ma	nas been established as if (some of) the amendments annexed to this report and listed below ade, since they have been considered to go beyond the disclosure as filed, as indicated in the ox (Rule 70.2(c)).
	☐ the desc	ription, pages
	☐ the clain☐ the draw	s, Nos. ings, sheets <i>t</i> iigs
	☐ the sequ	ence listing (specify):
	□ any table	e(s) related to sequence listing (specify):
	* If item 4	applies, some or all of these sheets may be marked "supporteded "

INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

International application No. PCT/EP2004/012851

Box No. V Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

1. Statement

Novelty (N) Yes: Claims 1-29

No: Claims -

Inventive step (IS) Yes: Claims 1-29

No: Claims

Industrial applicability (IA) Yes: Claims 1-29

No: Claims -

2. Citations and explanations (Rule 70.7):

see separate sheet

Re Item V

Reasoned statement with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

Reference is made to the following documents:

D4: PATENT ABSTRACTS OF JAPAN vol. 009, no. 188 (C-295), 3 August 1985 (1985-08-03) & JP 60 056086 A (HODOGAYA KAGAKU KOGYO KK), 1 April 1985 (1985-04-01)

Novelty (Article 33(2) PCT)

The compounds of claims 1-9 are not disclosed in the available prior art: the subject-matter of claims 1-9, 22-26 and 29 is therefore novel.

The single vessel method of claims 10-21 is not disclosed in the available prior art, whereby the subject-matter of these claims is novel.

The available prior art does not disclose electrodeposition of copper on printed circuit boards or on semiconductor substrates according to claims 27 and 28, whose subject-matter is therefore also novel.

Inventive step (Article 33(3) PCT)

D4, which is considered to represent the closest prior art, describes phenazinium derivatives useful as additives to improve copper-electrodeposition from plating baths.

In the light of the teachings of D4 the problem underlying and solved (see comparative examples of the description) by the present application can be seen in the provision of an improved process for copper-electrodeposition.

The use of the novel compounds, or of the purified compounds (not as mixtures), according to the present application to solve this problem is not suggested by the

INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY (SEPARATE SHEET)

International application No.

PCT/EP2004/012851

teachings of the available prior art, and involves therefore an inventive step.

Industrial applicability (Article 33(4) PCT)

The compounds, processes and compositions of claims 1-29 can be used in the electrochemical industry.

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Claims:

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5 1. Halogenated or pseudohalogenated monomeric phenazinium compounds of a purity of at least 85 mole-% having the following general chemical formula:

$$R^{8}$$
 R^{7}
 R^{7}
 R^{7}
 R^{6}
 R^{5}
 R^{4}
 R^{4}

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wherein

R¹, R², R⁴, R⁶, R⁷, R⁷, R⁸ and R⁹ are selected independently of each other from a group comprising hydrogen, halogen, amino, aminoalkyl, hydroxy, cyano, thiocyanate, isothiocyanate, cyanate, isocyanate, mercapto, carboxy, the salt thereof, carbonic acid ester, sulfo, the salt thereof, sulfoester, lower alkyl, unsubstituted aryl, substituted aryl, heteroaryl and alicyclic heteroradicals.

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R⁵ is selected from a group comprising lower alkyl, unsubstituted aryl, substituted aryl and heteroaryl.

20

X is a halogen or a pseudohalogen and

A is an acid anion.

* < in sert text of page 28 a>

25 2.

The phenazinium compounds according to claim 1, characterized in that R¹, R², R³, R⁶, R⁷, R⁷", R⁸ and R⁹ are selected independently of each other from a group comprising hydrogen and lower alkyl.

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with the requirement that, if X is halogen, the phenazinium compounds are selected from the group, comprising

- i) 3-chloro-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium salt,
- ii) 3-bromo-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium salt, and
- iii) 3-bromo-7-N,N-diethylamino-5-phenyl-phenazinium salt.

- The phenazinium compounds according to claim 2, characterized in that lower alkyl is methyl or ethyl.
- The phenazinium compounds according to any one of the preceding claims,
 characterized in that R⁵ is aryl.
 - 5. The phenazinium compounds according to claim 4, characterized in that aryl is phenyl.
- 10 16. The phenazinium compounds according to any one of the preceding claims; characterized in that X is chlorine, bromine or thiocyanate:
 - The phenazinium compounds according to any one of the preceding claims, characterized in that they are selected from a group comprising
 - i) +3-chloro-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium-salt,
 ii) 3-bromo-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium-salt,
 iii) 3-bromo-7-N,N-diethylamino-5-phenyl-phenazinium-salt and
 iv)i 7-amino-2,8-dimethyl-3-thiocyanato-5-phenyl-phenazinium salt.
- The phenazinium compounds according to Flaim 7, characterized in that the salt is selected from a group comprising chloride, bromide, hydrogen sulfate and tetrafluoroborate.
- The phenazinium compounds according to any one offclaims 7 and 8, characterized in that they are selected from a group comprising
 - i) 3-chloro-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium chloride,
 - ii) 3-bromo-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium bromide.
 - iii) 3-bromo-7-N,N-diethylamino-5-phenyl-phenazinium bromide and
 - iv) 7-amino-2,8-dimethyl-3-thiocyanato-5-phenyl-phenazinium tetrafiluoroborate.

19. The phenazinium compounds according to any one of the preceding claims, obtainable according to the following method:

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 a) forming a diazonium compound by diazotization of a monomeric phenazinium compound comprising at least one primary amino group in the presence of mineral acid and diazotization means in a first reaction step,

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 reacting the diazonium compound in a second reaction step to the halogenated or pseudohalogenated monomeric phenazinium compound in the presence of mineral acid and halide or pseudohalide,

wherein the first and the second reaction steps are both run in one single vessel.

15 11.

A method of preparing the halogenated or pseudohalogenated monomeric phenazinium compounds accordance with one of claims 1—10, comprising the following reaction steps:

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 a) forming a diazonium compound by diazotization of a monomeric phenazinium compound comprising at least one primary amino group in the presence of mineral acid and diazotization means in a first reaction step,

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b) reacting the diazonium compound in a second reaction step to the halogenated or pseudohalogenated monomeric phenazinium compound in the presence of mineral acid and halide or pseudohalide,

characterized in that the first and the second reaction steps are both run in one single vessel.

* (insert text of page 30 a)

The method according to claim 17, characterized in that the mineral acid is selected from a group comprising hydrogen halides, sulfuric acid, tetrafluoroboric acid, hexafluorophosphoric acid, phosphoric acid and the mixtures thereof with the proviso that no hydrogen halide is used in the preparation of the pseudohalogenated monomeric phenazinium compounds.

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of a purity of at least 85 mole-% having the following general chemical formula:

$$R^{8}$$
 R^{9}
 R^{1}
 R^{2}
 R^{7}
 R^{7}
 R^{6}
 R^{5}
 R^{4}

wherein

R¹, R², R⁴, R⁶, R⁷, R⁷", R⁸ and R⁹ are selected independently of each other from a group comprising hydrogen, halogen, amino, aminoalkyl, hydroxy, cyano, thiocyanate, isothiocyanate, cyanate, isocyanate, mercapto, carboxy, the salt thereof, carbonic acid ester, sulfo, the salt thereof, sulfoester, lower alkyl, unsubstituted aryl, substituted aryl, heteroaryl and alicyclic heteroradicals,

15 R⁵ is selected from a group comprising lower alkyl, unsubstituted aryl, substituted aryl and heteroaryl,

X is a halogen or a pseudo halogen and

20 A is an acid anion

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12. The method according to any one of claims 14 and 12, characterized in that the diazotization means is metal nitrite or nitrosylsulfuric acid.

The method according to claim 13, characterized in that the metal nitrite is sodium nitrite.

18. The method according to any one of claims 11 - 14, characterized in that the halide is added in the form of hydrogen halide and/or in the form of a metal halide.

18. The method according to claim 18, characterized in that the metal halide is selected from a group comprising transition metal halides.

The method according to claim 16, characterized in that the transition metal halides are selected from a group comprising copper(I) halides, copper(II) halides, nickel(II) halides and iron(II) halides.

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18. The method according to any one of claims 11 - 11, characterized in that the halide is selected from a group comprising fluoride, chloride and bromide.

18. The method according to any one of claims 1/1 – 17, characterized in that the pseudohalide is selected from a group comprising cyanate (–OCN), thiocyanate (–SCN), isocyanate (–NCO) and isothiocyanate (–NCS).

20. The method according to any one of claims 1/1 – 19, comprising the following method steps:

i) mixing the monomeric phenazinium compounds, comprising at least one primary amino group, with the mineral acid,

- ii) next heating to a temperature above room temperature,
- iii) adding the halide or pseudohalide
- iv) adding the diazotization means.

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- The method according to claim 20, characterized in that, in the preparation of the halogenated monomeric phenazinium compounds, the mineral acid is hydrogen halide and step iii) is eliminated.
- The method according to any one of claims 17-21, characterized in that the following halogenated or pseudohalogenated monomeric phenazinium compounds are prepared:
 - a) 3-chloro-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium chloride,
- b) 3-bromo-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium bromide,
 - c) 3-bromo-7-N,N-diethylamino-5-phenyl-phenazinium bromide,
 - d) 7-amino-2,8-dimethyl-3-thiocyanato-5-phenyl-phenazinium tetraflluoroborate.
- 22
 23. An acidic bath for electrolytically depositing a copper deposit, containing at least one halogenated or pseudohalogenated monomeric phenazinium compound in accordance with any one of claims 1 1%.
- 23
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 24. The bath according to claim 28, characterized in that the phenazinium compounds are contained in a concentration of from 0.00005 0.1 g/l.
 - 22 23
 28. The bath according to one any of claims 28 and 24, characterized in that it additionally contains compounds selected from a group comprising nitrogencontaining sulfur compounds and polymeric nitrogen compounds.
 - 25 24
 26. The bath according to claim 25, characterized in that the concentration of the nitrogen-containing sulfur compounds and the polymeric nitrogen compounds contained together in the bath is from 0.0001 0.50 g/l.
 - 22 25
 27. Use of the bath according to any one of claims 23 26 for depositing a mirror bright, leveled copper deposit for the purpose of producing decorative surfaces.

28. Use of the bath according to any one of claims 23 - 26 for forming a copper-deposit on printed circuit board material. * in sert text of page 33a>

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29. Use of the bath according to any one of claims 23 - 26 for forming a copperdeposit on semiconductor substrates. * < in sev t text of page 336>

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20. A method of electrolytically depositing a copper deposit onto a workpiece by which the workpiece and at least one anode are contacted with the bath according to one of claims 28 – 26, and a flow of electric current is generated between the workpiece and the anodes.

page 33a

27. Use of an acid bath, containing at least one halogenated or pseudohalogenated monomeric phenazinium compound of a purity of at least 85 mole-% having the following general chemical formula:

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$$R^8$$
 R^9
 R^1
 R^2
 R^7
 R^7
 R^6
 R^5
 R^4

wherein

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R¹, R², R⁴, R⁶, R⁷, R⁷", R⁸ and R⁹ are selected independently of each other from a group comprising hydrogen, halogen, amino, aminoalkyl, hydroxy, cyano, thiocyanate, isothiocyanate, cyanate, isocyanate, mercapto, carboxy, the salt thereof, carbonic acid ester, sulfo, the salt thereof, sulfoester, lower alkyl, unsubstituted aryl, substituted aryl, heteroaryl and alicyclic heteroradicals,

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R⁵ is selected from a group comprising lower alkyl, unsubstituted aryl, substituted aryl and heteroaryl,

X is a halogen or a pseudo halogen and

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A is an acid anion,

for electrolytically depositing a copper deposit on printed circuit board material.

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page 336

28. Use of an acid bath, containing at least one halogenated or pseudohalogenated monomeric phenazinium compound of a purity of at least 85 mole-% having the following general chemical formula:

$$\mathbb{R}^{8}$$
 \mathbb{R}^{9}
 \mathbb{R}^{1}
 \mathbb{R}^{2}
 \mathbb{R}^{7}
 \mathbb{R}^{7}
 \mathbb{R}^{7}
 \mathbb{R}^{1}
 \mathbb{R}^{2}
 \mathbb{R}^{4}
 \mathbb{R}^{4}

wherein

R¹, R², R⁴, R⁶, R⁷, R⁷, R⁸ and R⁹ are selected independently of each other from a group comprising hydrogen, halogen, amino, aminoalkyl, hydroxy, cyano, thiocyanate, isothiocyanate, cyanate, isocyanate, mercapto, carboxy, the salt thereof, carbonic acid ester, sulfo, the salt thereof, sulfoester, lower alkyl, unsubstituted aryl, substituted aryl, heteroaryl and alicyclic heteroradicals,

R⁵ is selected from a group comprising lower alkyl, unsubstituted aryl, substituted aryl and heteroaryl,

X is a halogen or a pseudo halogen and

20 A is an acid anion,

for electrolytically depositing a copper deposit on semiconductor substrates.

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wherein

R¹, R², R⁴, R⁶, R⁷′, R⁷″, R⁸ and R⁹ are selected independently of each other from a group comprising hydrogen, halogen, amino, aminoalkyl, hydroxy, cyano, thiocyanate, isothiocyanate, cyanate, isocyanate, mercapto, carboxy (COO¯), the salt thereof, carbonic acid ester (COOR), sulfo (SO₃¯), the salt thereof, sulfoester (SO₃R), lower alkyl, unsubstituted aryl, substituted aryl, heteroaryl and alicyclic heteroradicals,

R⁵ is selected from a group comprising lower alkyl, unsubstituted aryl, substituted aryl and heteroaryl,

X is a halogen or a pseudohalogen and

A is an acid anion,

* < insert text of page [a >

The phenazinium compounds of a purity of x mole-% as mentioned herein above, herein after and in the claims refer to a mixture of the phenazinium compounds and impurities, with the phenazinium compounds being contained in the mixture at a concentration of x mole-% and the impurities at a concentration of 100 - x mole-%.

The term lower alkyl as mentioned herein above and herein after and in the claims preferably refers to C_1 - to C_4 -alkyl, meaning to methyl, ethyl, n-propyl, iso-propyl, n-butyl, iso-butyl and tert-butyl. By substituted alkyl as mentioned herein above, herein after and in the claims, sulfo- or carbonic acid-substituted alkyl is preferably meant.

Aryl as mentioned herein above or herein after and in the claims preferably refers to phenyl or polycyclic aromates such as naphthyl-1 and naphthyl-2, wherein these residues may be unsubstituted or substituted respectively. If these residues are substituted, they are more specifically substituted by alkyl, preferably by lower alkyl, halogen, hydroxy, amino, wherein amino is NH₂, NHR or NR'R", wherein R, R' and R" in turn can be lower alkyl, cyano, thiocyanate and mercapto. Phenyl may more specifically be substituted at a 2-, 4- and 6-position.

with the requirement that, if X is halogen, the phenazinium compounds are selected from the group, comprising

- i) 3-chloro-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium salt,
- ii) 3-bromo-7-N,N-dimethylamino-2-methyl-5-phenyl-phenazinium salt, and
- iii) 3-bromo-7-N,N-diethylamino-5-phenyl-phenazinium salt.